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Synthesis of Le^x and Le^y Oligosaccharides with Azido-Type Spacer-Arms. Comparison of 3- and 4-Methoxybenzyl Groups as Key Temporary Protective Groups

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SYNTHESIS OF LE^x AND LE^y OLIGOSACCHARIDES WITH AZIDO-TYPE SPACER-ARMS. COMPARISON OF 3- AND 4-METHOXYBENZYL GROUPS AS KEY TEMPORARY PROTECTIVE GROUPS¹

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ABSTRACT

5-Azido-3-oxa-1-pentanol was prepared from 2-(2-chloroethoxy)ethanol and used as a spacer in the chemical synthesis of the trisaccharide β -D-Gal-(1 \rightarrow 4)-[α -L-Fuc-(1 \rightarrow 3)]-GlcNAc and the tetrasaccharide α -L-Fuc- α -(1 \rightarrow 2)- β -D-Gal-(1 \rightarrow 4)-[α -L-Fuc-(1 \rightarrow 3)]-GlcNAc that represent the epitopes defining the human blood groups Le^x and Le^y. The classical 4-methoxybenzyl group and the remarkably acid-stable 3-methoxybenzyl group were compared as temporary protective groups for position 3 at the glucosamine unit to circumvent the problems associated with the simultaneous presence of allyl and azido groups. The resulting oligosaccharides were coupled to proteins with high efficiency.

INTRODUCTION

Spacer-arm derivatives of oligosaccharides are frequently used for coupling to proteins. Several neoglycoproteins have been prepared in this way as immunogens for the production of carbohydrate-specific antibodies^{2,3} in animals or even as possible therapeutic tools for humans.^{4,5} However, after a multistep synthesis, very often a

precious oligosaccharide is coupled to a protein with a yield not exceeding 20%, based on the oligosaccharide.

We⁶ and others^{7,8} recently developed the use of 5-azido-3-oxa-1-pentanol for the preparation of spacer-armed oligosaccharides. The reduction of the azido group to an amine group followed by derivatization as a maleimide allowed almost quantitative coupling between the oligosaccharide and protein by using only a slight molar excess of the oligosaccharide. Other thiophilic groups introduced into the spacer through the terminal amino function were also coupled efficiently.

In an attempt to prepare the Le^x and Le^y haptens using this spacer, the classical allyl temporary protective group strategy failed owing to the problems associated with the selective removal of the allyl group in the presence of an azido function. In the present paper, we describe the synthesis of type 2 Lewis oligosaccharides and compare the use of 4-methoxybenzyl and the more acid resistant 3-methoxybenzyl group⁹ in the place of allyl.

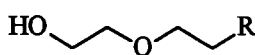
RESULTS AND DISCUSSION

2-(2-Chloroethoxy)ethanol (**1**) could be transformed directly by the action of sodium azide, tetrabutylammonium iodide and dicyclohexano-18-crown-6 in butanone into 5-azido-3-oxa-1-pentanol (**2**) in excellent yield. The reaction of 2-deoxy-2-acetamido-3,4,6-tri-*O*- α -D-glucopyranosyl chloride with **2** in the presence of mercury(II) cyanide proceeds smoothly to give crystalline glucosaminide **3** that was deacetylated and transformed into a key benzylidenated intermediate **4**, isolated by crystallization.

In previous syntheses of Le^x and Le^y, different strategies¹⁰ were employed but most use the allyl group as a temporary protective group¹¹ for position 3. In our case, the presence of allyl and azido groups in the same molecule caused attempts at selective removal of the allyl group¹² to give low yields and complex product mixtures. In search of a better protecting group, we first tried the 4-methoxybenzyl group as previously reported.¹³ Preliminary results pointed to the advantage of a more acid stable group in our sequence. We decided to assess the performance of both the 4- and the 3-methoxybenzyl groups during the syntheses of type 2 Lewis oligosaccharides.

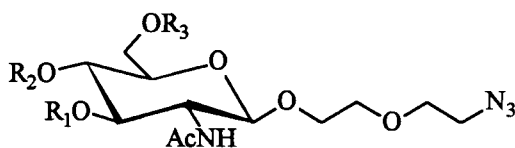
The benzylidene intermediate **4** was methoxybenzylated using the corresponding methoxybenzyl bromides¹⁴ in the presence of barium oxide-barium hydroxide¹⁵ in DMF.

Both crystalline derivatives **5** and **6** were then treated with sodium cyanoborohydride and HCl/ether in tetrahydrofuran¹⁶ to afford the 6-*O*-benzyl derivatives **7** and **8**. The yields were similar in the two cases, but for the 4-methoxybenzyl group, the acidity had to be controlled very carefully. The structures of **7** and **8** were ascertained by ¹³C NMR spectroscopy (see Table 1); the signals corresponding to C-6 were almost unaffected by substitution (**7**, 68.5 ppm → 68.7 ppm and **8**, 68.5 ppm → 68.6 ppm) while those corresponding to C-4 were shielded (**7**, 82.1 ppm → 75.5 ppm; **8**, 82.6 ppm → 75.2 ppm).



1 R = Cl

2 R = N₃



3 R₁ = R₂ = R₃ = Ac

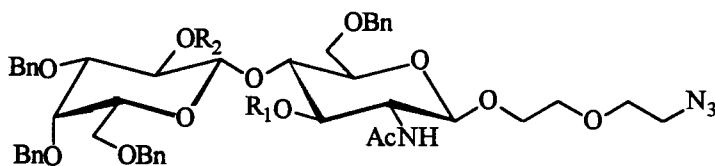
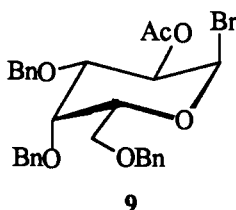
4 R₁ = H, R₂, R₃ = PhCH

5 R₁ = mMBn, R₂, R₃ = PhCH

6 R₁ = pMBn, R₂, R₃ = PhCH

7 R₁ = mMBn, R₂ = H, R₃ = Bn

8 R₁ = pMBn, R₂ = H, R₃ = Bn



10 R₁ = mMBn, R₂ = Ac

11 R₁ = pMBn, R₂ = Ac

12 R₁ = H, R₂ = Ac

13 R₁ = R₂ = H

Galactosylations of **7** and **8** were performed with 2-*O*-acetyl-3,4,6-tri-*O*-benzyl- α -D-galactopyranosyl bromide^{11e} in dichloromethane in the presence of silver triflate.¹⁷

Table 1. ^{13}C NMR spectral data^a for compounds **3-14**

compd	Glucosamine						Galactose					
	C1	C2	C3	C4	C5	C6	C1'	C2'	C3'	C4'	C5'	C6'
3	100.9	54.0	72.4	70.8	71.5	62.0						
4	101.5	56.1	70.4	81.2	66.0	67.9						
5	100.9	56.2	77.6	82.1	65.8	68.5						
6	100.8	57.2	76.4	82.6	66.0	68.5						
7	101.7	55.6	83.0	75.1	72.3	68.7						
8	101.2	56.2	81.5	75.2	72.6	68.6						
10	101.3	52.7	79.1	76.0	75.2	68.5	100.9	72.2	80.7	73.2	73.8	69.5
11	101.2	51.4	77.5	75.1	75.1	68.4	100.6	72.2	80.6	73.4	73.9	69.0
12	101.5	55.8	73.1	80.4	74.1	68.5	100.7	71.1	80.0	72.1	73.6	68.8
13	101.3	56.5	73.1	82.4	74.0	68.2	104.4	71.1	82.0	73.2	73.8	68.8
14	99.6	56.3	73.3	72.2	75.8	68.5	99.8	72.0	80.3	72.9	72.4	68.2
Fuc	97.0	75.2	79.6	78.2	66.6	16.2						
15	99.3	59.6	73.6	73.1	75.4	68.4	100.2	72.8	83.8	73.3	72.3	67.8
Fuc	98.0	75.3	79.8	78.2	66.7	16.2	97.8	75.7	78.9	78.0	66.4	16.1

a. Chemical shifts for protective groups are as follows: CH_2N_3 50.1-50.8 ppm; 3-methoxybenzyl 140.5, 113.5, 160.3, 113.9, 129.7 and 120.4 ppm; 4-methoxybenzyl 134.7, 131.4, 114.2 and 159.8 ppm; benzylidene 101.2, 127.4-129.3 ppm; Bn 72.0-74.6 and 127.4-129.3 ppm; CH_3CONH 23.2-23.4 and 170.1-170.4 ppm; CH_3CO 20.9 and 169.3 ppm.

These conditions led to disaccharide **10** in 60% yield. However, disaccharide **11** was obtained¹⁸ in acceptable yield only when *N,N*-diisopropylethylamine was added to prevent the acid cleavage of the 4-methoxybenzyl group. The structures of disaccharides **10** and **11** were confirmed by the presence in their ^1H NMR spectra of doublets at 4.55 ppm (*J* 7.7 Hz) and 4.48 ppm (*J* 7.0 Hz), respectively.

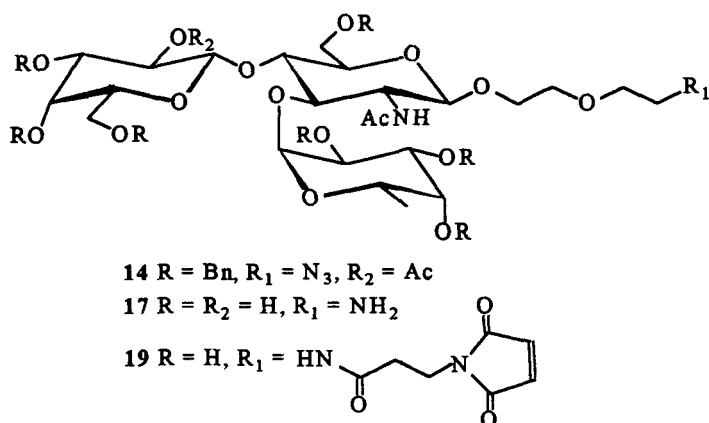
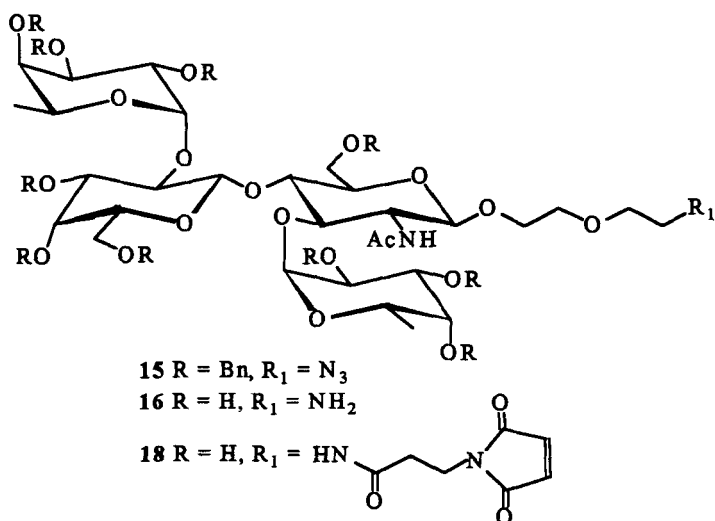
In previous reactions, even if the yields were only slightly different, the two groups displayed distinctive behavior under acidic conditions. The 3-methoxybenzyl group is as stable as a benzyl or allyl group, while the 4-methoxybenzyl group was as acid sensitive as an acetal.

Oxidative removal of *O*-methoxybenzyl groups from disaccharides **10** and **11** with DDQ in dichloromethane-water proceeded better for **11** as expected. It was also possible to remove the 3-methoxybenzyl group from **10** albeit in a lower yield as these conditions led to partial removal of some benzyl groups on the galactose unit. The remarkable acid stability of the latter group deserves future study to improve the selectivity of deprotection. The structure of **12** was established based upon the absence of the signal corresponding to the methoxybenzyl group and by the shielding of the C-3 signal (79.1 ppm → 73.1 ppm) in the ¹³C NMR spectrum.

Fucosylation of **8** was performed with tri-*O*-benzyl- α -L-fucopyranosyl bromide^{19,20} as the donor using the halide-ion catalyzed reaction²¹ to give the α -L-linked trisaccharide **14** in 82% yield. The disaccharide **12** was first deacetylated and then the acceptor **13** was di- α -L-fucosylated in the same way to afford the Lewis^y tetrasaccharide **15** in 60% yield.

After hydrogenolysis, the spacer amino group of the corresponding Le^x and Le^y free oligosaccharides (**16** and **17**) reacted with the *N*-hydroxysuccinimide derivative of β -maleimidopropionic acid following the procedure previously described for model oligosaccharides²² to give the corresponding β -maleimidopropionamide derivatives **18** and **19**. The ¹H NMR spectra showed complete transformation of the free amino group into the corresponding amides (δ 3.14→3.26). Small amounts of β -maleimidopropionic acid, that were very difficult to remove, were detected by the presence of triplets at 2.44 ppm in the ¹H NMR spectra. Compounds **18** and **19** are stable in aqueous solution at pH < 6.5 but hydrolysed slowly at pHs above this value. However, the reactions with the BSA thiol-groups were several times faster than hydrolysis at pH 7.2 and proceeded smoothly with only one equivalent of oligosaccharide per SH-group.

Incorporations ranging from 13 to 20 mol of oligosaccharide per mol of BSA were usually obtained that represent yields between 50-80 % based on the oligosaccharide. The presence of the β -maleimidopropionic acid did not affect the rate of linkage formation to the oligosaccharide. The use of this and other neoglycoproteins as immunogens for the preparation of monoclonal antibodies is now in progress.



EXPERIMENTAL

General procedures. Optical rotations were measured at 25 °C with a POLAMAT A automatic polarimeter, using a 5 cm 5 mL cell. NMR spectra were recorded at 25 °C with a BRUKER AC-250F spectrometer. Chemical shifts (δ) are given in ppm relative to the signal for internal tetramethylsilane for ^1H NMR spectra and are referenced to the central line of CDCl_3 , δ 77.03, for ^{13}C NMR spectra. Assignments were made on the basis of homonuclear and heteronuclear correlation experiments. The following notations are used for identification of monosaccharide units

in the NMR spectra: ' for Gal; f and f' for the Fuc unit linked to GlcNAc or Gal, respectively.

All compounds were purified by column chromatography on Kieselgel 60 (Fluka, < 230 mesh ASTM) and fractions were monitored by TLC on Kieselgel 60 F₂₅₄ (Merck). Detection was effected by charring with sulfuric acid after examination under UV light. Evaporations were conducted under reduced pressure at 40 °C (bath).

5-Azido-3-oxa-1-pentanol (2).²³ To a solution of 2-(2-chloroethoxy)ethanol (1) (5 mL, 47 mmol) in 2-butanone (25 mL) was added sodium azide (4.5 g, 69 mmol), tetrabutylammonium iodide (2.5 g, 6 mmol) and dicyclohexano-18-crown-6 (10 mg). The mixture was refluxed at 90 °C for 24 h, when ¹³C NMR spectroscopy of the supernatant liquid showed the absence of a signal at δ 42.7 and the presence of a strong signal at δ 50.0 ppm. The mixture was filtered, the solids were rinsed with acetone and the combined solutions were concentrated. Distillation of the residue gave, at 60-80 °C and 0.2 mbar, compound 2 (4.8 g, 78.6%): ¹H NMR (CDCl₃) δ 3.71-3.6 (m, 6H, CH₂O) and 3.41 (t, 2H, CH₂N₃); ¹³C NMR (CDCl₃) δ 72.37 and 69.84 (CH₂O), 61.56 (CH₂OH) and 50.62 (CH₂N₃).

5-Azido-3-oxapentyl 2-Acetamido-3,4,6-tri-O-acetyl-2-deoxy-β-D-glucopyranoside²⁴ (3). A solution of 2-acetamido-2-deoxy-3,4,6-tri-O-acetyl-α-D-glucopyranosyl chloride (5 g, 15.1 mmol) and compound 2 (2.33 g, 17.7 mmol) in anhydrous dichloromethane (20 mL) containing 0.4 nm molecular sieves (5 g) and drierite (5 g) was stirred for 1 h under a nitrogen atmosphere. Mercury(II) cyanide (3.45 g, 13.6 mmol) was added and the mixture was stirred for 48 h. Then, the mixture was diluted with dichloromethane (50 mL), filtered through Celite and the filtrate washed with aqueous 10% potassium iodide (40 mL), saturated aqueous sodium hydrogen carbonate (40 mL), water, dried (Na₂SO₄), filtered and concentrated. Crystallisation from ethyl acetate/diethyl ether afforded 4 (4.84 g, 77.5%): mp 100-102 °C; [α]_D +54.0° (c 1.0, chloroform); R_F 0.54 (dichloromethane/acetone, 4:1 v/v); ¹H NMR (CDCl₃) δ 6.35 (δ, 1H, J = 10.2 Hz, NH), 5.21-5.28 (m, 2H, H-3, 4), 4.78 (d, 1H, J_{1,2} = 7.6 Hz, H-1), 4.35 (dd, J_{6a,6b} = 10.3 Hz, J_{5,6a} = 5.3 Hz, 1H, H-6a), 4.15 (dd, 1H, H-6b), 3.95 (m, 1H, H-2), 3.75 (m, 1H, H-5), 3.65 (m, 6H, CH₂ spacer), 3.42 (t, 2H, CH₂N₃) and 2.1-1.95 (3s, 12H, CH₃CON and CH₃COO).

Anal. Calcd. for $C_{18}H_{28}O_{10}N_4$ (460.64): C, 46.95; H, 6.12, N, 12.17. Found: C, 47.26; H, 5.92; N 12.11.

5-Azido-3-oxapentyl 2-Acetamido-4,6-O-benzylidene-2-deoxy- β -D-glucopyranoside (4). To a solution of compound **4** (3.8 g, 8.2 mmol) in dry methanol (50 mL), was added sodium methoxide (0.1 M) to pH 9. After 1 h, the reaction was neutralised with Dowex-50 (H^+) resin, filtered and concentrated. A solution of the residue (2.56 g, 7.6 mmol) and anhydrous zinc chloride (2.7 g, 7.6 mmol) in benzaldehyde (16 mL, 146.2 mmol) was stirred for 24 h. The mixture was then poured into ice-water (250 mL) and hexane (100 mL) with vigorous stirring. The solid was filtered, rinsed thoroughly with hexane and crystallised from absolute ethanol to afford **4** (2.6 g, 65%): mp 230-232 °C; $[\alpha]_D -72.0^\circ$ (c 1, DMSO); R_F 0.27 (dichloromethane/acetone, 4:1 v/v); 1H NMR (DMSO- d_6) δ 7.84 (d, 1H, $J = 8.5$ Hz, NH), 7.47-7.37 (m, 5H, Ph), 5.61 (s, 1H, PhCH), 5.32 (d, 1H, $J_{3,OH} = 5.4$ Hz, OH-3), 4.54 (d, 1H, $J_{1,2} = 7.9$ Hz, H-1), 4.25 (dd, 1H, $J_{5,6a} = 4.6$ Hz, $J_{6a,6b} = 10.0$ Hz, H-6a), 3.74 (m, 1H, H-6b), 3.58 (m, 2H, H-2,3), 3.41 (m, 3H, H-4, CH_2N_3), 3.34 (m, 1H, H-5) and 1.84 (s, 3H, CH_3CON).

Anal. Calcd for $C_{19}H_{26}O_7N_4$ (422.43): C, 54.01; H, 6.20; N, 13.26. Found: C, 53.65; H, 6.58; N, 13.33.

5-Azido-3-oxapentyl 2-Acetamido-4,6-O-benzylidene-2-deoxy-3-O-(3-methoxybenzyl)- β -D-glucopyranoside (5). To a solution of **4** (2.5 g, 5.35 mmol), barium oxide (4.1 g, 26.75 mmol) and $Ba(OH)_2 \cdot 8H_2O$ (740 mg, 2.3 mmol) in N,N -dimethylformamide (20 mL) was added 3-methoxybenzyl bromide (1.83 mL, 12.6 mmol) and the mixture was stirred for 15 min at rt. Dichloromethane (50 mL) was then added and the resulting mixture was refluxed for 1 h and then filtered through Celite. The filtrate was washed with 2% hydrochloric acid (25 mL), saturated aqueous sodium hydrogen carbonate (25 mL) and water (25 mL), then dried and concentrated. After the addition of toluene (10 mL) the solid was filtered, rinsed thoroughly with toluene and recrystallized from ethyl acetate to afford **5** (2.6 g, 85%): mp 207-209 °C; $[\alpha]_D -15.6^\circ$ (c 0.78, chloroform); R_F 0.69 (dichloromethane/ acetone 5:1 v/v); 1H NMR ($CDCl_3$) δ 7.5-7.2 (m, 6H, Ph), 6.65 (m, 3H, H-2, 4, 6 PhOMe), 6.35 (d, 1H, $J = 8.5$ Hz, NH), 5.65 (s, 1H, PhCH), 4.95 (d, 1H, $J = 8.3$ Hz, H-1), 4.70 (AB, CH_2PhOCH_3), 4.36 (dd, 1H, $J_{5,6a} = 4.1$ Hz, $J_{6a,6b} = 10.25$ Hz, H-6a), 4.10 (t, $J_{2,3} = 9.3$ Hz, 1H, H-3), 3.95 (m, 1H, H-2), 3.73 (m, 5H, H-4, H-6b, CH_3O), 3.50 (m, 1H, H-5), 3.38 (t, 2H, $J = 4.9$ Hz, CH_2N_3) and 1.95 (s, 3H, CH_3CON).

Anal. Calcd for C₂₇H₃₄O₈N₄ (542.59): C, 59.77; H, 6.32; N, 10.33. Found: C, 59.30; H, 6.58; N, 10.20.

5-Azido-3-oxapentyl 2-Acetamido-4,6-O-benzylidene-2-deoxy-3-O-(4-methoxybenzyl)-β-D-glucopyranoside (6). Compound 6 was obtained from 4 as described above for the preparation of 5: yield 83.2%; mp 207-209 °C; [α]_D -19.8° (c 0.8, chloroform); R_F 0.69 (dichloromethane/acetone 5:1 v/v); ¹H NMR (CDCl₃) 7.55-7.45 (m, 5H, Ph), 7.30 (d, 2H, H-2, 6 PhOCH₃), 6.70 (d, 2H, H-3, 5 PhOCH₃) 5.75 (d, J = 8.5 Hz, 1H, NH), 5.58 (s, 1H, CHPh), 4.95 (d, J = 8.3 Hz, 1H, H-1), 4.70 (AB pattern, CH₂PhOCH₃), 4.36 (dd, J_{5,6a} = 4.1 Hz, J_{6a,6b} = 10.25 Hz, 1H, H-6a), 4.15 (t, J_{2,3} = 9.3 Hz, 1H, H-3), 3.95 (m, 1H, H-2), 3.70 (m, 5H, H-4, H-6b, CH₃O), 3.50 (m, 1H, H-5), 3.40 (t, J = 4.9 Hz, 2H, CH₂N₃) and 1.95 (s, 3H, CH₃CON).

Anal. Calcd for C₂₇H₃₄O₈N₄ (542.59): C, 59.77; H, 6.32; N, 10.33. Found: C, 59.15; H, 6.31; N, 10.27.

5-Azido-3-oxapentyl 2-Acetamido-6-O-benzyl-2-deoxy-3-O-(3-methoxybenzyl)-β-D-glucopyranoside (7). A mixture of compound 5 (1g, 1.7 mmol), sodium cyanoborohydride (1.07 g, 17 mmol) and 0.3 nm molecular sieves (2 g) in dry tetrahydrofuran (20 mL) was stirred for 15 min at rt. Then the mixture was cooled to 0 °C and dry diethyl ether saturated with hydrogen chloride was added at 0 °C until the evolution of gas stopped. The cooling bath was removed and the reaction mixture was further stirred for 20 min. Cold water (5 mL) was added, then the suspension was diluted with dichloromethane (50 mL) and filtered through Celite. The filtrate was washed with 1 % aqueous potassium permanganate (3 x 20 mL), saturated aqueous sodium hydrogen carbonate (20 mL) and water (20 mL), then dried and concentrated. Column chromatography (dichloromethane/acetone 5:1 v/v) of the residue afforded 6 (600 mg, 60 %) as a colorless solid. An analytical sample was obtained by recrystallization from ethyl acetate: mp 96-98 °C; [α]_D -10.9° (c 6.2, chloroform); R_F 0.48 (dichloromethane/acetone 4:1 v/v); ¹H NMR (C₆D₆) δ 7.45-7.35 (m, 6H, Ph), 6.90 (m, 3H, H-2,4,6 PhOMe), 6.62 (d, 1H, J = 7.7 Hz, NH), 4.97 (AB pattern, 2H, CH₂Ph), 4.69 (d, 1H, J_{1,2} = 8.3 Hz, H-1), 4.41 (s, 2H, PhCH₂), 4.15 (m, 1H, H-2), 3.88 (m, 3H, H-3, 4, 6a), 3.70 (m, 1H, H-5), 3.52 (s, 3H, OCH₃), 3.45 (m, 1H, H-6b), 3.23 (t, 2H, H-CH₂CH₂N₃), 2.87 (t, 2H, J = 4.9 Hz, CH₂N₃) and 1.82 (s, 3H, CH₃CON).

Anal. Calcd for C₂₇H₃₆O₈N₄ (544.60): C, 59.54; H, 6.66; N, 10.29. Found: C, 58.98; H, 6.71; N, 10.08.

5-Azido-3-oxapentyl 2-Acetamido-6-O-benzyl-2-deoxy-3-O-(4-methoxybenzyl)- β -D-glucopyranoside (8). A mixture of compound **6** (1g, 1.7 mmol), sodium cyanoborohydride (1.07 g, 17 mmol) and 0.3 nm molecular sieves (2 g) in dry tetrahydrofuran (20 mL) was stirred for 15 min at rt. Then the mixture was cooled to 0 °C and dry diethyl ether saturated with hydrogen chloride was added at 0 °C until the evolution of gas stopped and the solution was stirred for 10 min at 0 °C. Cold water (5 mL) was added, and the suspension was diluted with dichloromethane (50 mL) and filtered through Celite. The filtrate was washed with 1% aqueous potassium permanganate (3 x 20mL) saturated aqueous sodium hydrogen carbonate (20 mL) and water (20 mL), dried and concentrated. Column chromatography (dichloromethane/acetone 5:1 v/v) of the residue afforded **8** (561 mg, 56%) as a colorless solid. An analytical sample was obtained by recrystallization from ethyl acetate: mp 88-90 °C; $[\alpha]_D -8.18^\circ$ (*c* 8.8, chloroform); R_F 0.48 (dichloromethane/acetone 4:1 v/v); 1H NMR (C_6D_6) δ 7.45-7.38 (m, 5H, Ph), 7.30 (d, 2H, H-2, H-6 PhOCH₃), 6.80 (d, 2H, H-3, 5 PhOCH₃) 5.92 (d, *J* = 7.7 Hz, 1H, NH), 4.87 (AB, 2H, CH₂Ph), 4.65 (d, *J*_{1,2} = 8.3 Hz, 1H, H-1), 4.41 (s, 2H, PhCH₂), 3.82(m, 4H, H-2, 3, 4, 6a), 3.65 (m, 1H, H-5), 3.35 (s, 3H, OCH₃), 3.35 (m, 2H, H-6b), 3.23 (t, 2H, CH₂CH₂N₃), 2.87 (t, *J* = 4.9 Hz, 2H, CH₂N₃) and 1.82 (s, 3H, CH₃CON).

Anal. Calcd for C₂₇H₃₆O₈N₄ (544.60): C, 59.54; H, 6.66; N, 10.29. Found: C, 59.78; H, 6.86; N, 10.16.

5-Azido-3-oxapentyl 2-Acetamido-4-O-(2-O-acetyl-3,4,6-tri-O-benzyl- β -D-galactopyranosyl)-6-O-benzyl-2-deoxy-3-(3-methoxybenzyl)- β -D-glucopyranoside (10). A solution of **6** (200 mg, 0.35 mmol) and silver triflate (300 mg, 1.17 mmol) in dry dichloromethane (3 mL) containing molecular sieves (1.5 g) was stirred under nitrogen for 15 min at rt and then a solution of the bromide **9** (647 mg, 1.17 mmol) in dichloromethane (3 mL) was added. After being stirred for 6 h under nitrogen at rt, the mixture was diluted with dichloromethane (10 mL) and filtered through Celite. The filtrate was washed with water (5 mL), saturated sodium hydrogen carbonate (5 mL) and water (5 mL), then dried and concentrated. Column chromatography (dichloromethane/acetone 8:1 v/v) of the residue afforded **10** as a syrup (210 mg, 60%): $[\alpha]_D -47.0^\circ$ (*c* 1.0, chloroform); R_F 0.65 (dichloromethane/acetone 4:1 v/v); 1H NMR (C_6D_6) δ 7.50-7.30 (m, 21H, Ph), 6.80 (m, 3H, H-2,4,6 PhOMe), 6.60 (d, *J* = 9.1 Hz, 1H,

NH), 5.72 (dd, $J_{1,2'} = 7.7$ Hz, $J_{2',3'} = 10.1$ Hz, 1H, H-2'), 4.78 (d, $J_{1,2} = 6.2$ Hz, 1H, H-1), 4.55 (d, 1H, H-1'), 4.40 (m, 1H, H-2), 4.20 (m, 1H, H-4), 3.87 (m, 4H, H-6a, 6a', 3, 4'), 3.61 (m, 1H, H-5), 3.54 (s, 3H, OCH₃), 3.37 (m, 2H, H-6b, 5'), 2.90 (t, $J = 4.9$ Hz, 2H, CH₂N₃), 1.91 (s, 3H, CH₃CON) and 1.87 (s, 3H, Ac).

Anal. Calcd for C₅₆H₆₆O₁₄N₄ (1019.16): C, 66.00; H, 6.53; N, 5.50. Found: C, 66.17; H, 6.59; N, 5.31.

5-Azido-3-oxapentyl 2-Acetamido-4-O-(2-O-acetyl-3,4,6-tri-O-benzyl-β-D-galactopyranosyl)-6-O-benzyl-2-deoxy-3-(4-methoxybenzyl)-β-D-glucopyranoside (11). A solution of **6** (100 mg, 0.18 mmol), silver triflate (162 mg, 0.63 mmol) and *N,N*-di-isopropylethylamine (32 μL, 0.18 mmol) in dry dichloromethane (3 mL) containing 0.4 nm molecular sieves (1.5 g) was stirred under nitrogen for 15 min at rt and then a solution of bromide **9** (350 mg, 0.63 mmol) in dichloromethane (3 mL) was added. After being stirred for 6 h at rt, the mixture was diluted with dichloromethane (10 mL) and filtered through Celite. The filtrate was washed with water (5 mL), saturated sodium hydrogencarbonate (5 mL) and water (5 mL), then dried and concentrated. Column chromatography (dichloromethane/acetone 8:1 v/v) of the residue afforded **11** (106 mg, 57%) as a syrup; $[\alpha]_D -66.3^\circ$ (*c* 1.17, chloroform); R_F 0.65 (dichloromethane/acetone 4:1 v/v); ¹H NMR (C₆D₆) δ 7.50-7.40 (m, 20H, Ph), 7.25 (d, 2H, H-2, 6PhOCH₃), 6.75 (d, 2H, H-3,5 PhOCH₃) 6.21 (d, $J = 8.9$ Hz, 1H, NH), 5.80 (dd, $J_{1,2'} = 7.7$ Hz, $J_{2',3'} = 10.1$ Hz, 1H, H-2'), 4.85 (AB pattern, 2H, CH₂Ph), 4.80 (d, $J_{1,2} = 6.5$ Hz, 1H, H-1), 4.48 (d, 1H, H-1'), 4.48 (m, 1H, H-2), 4.35 (m, 1H, H-4), 3.95 (m, 2H, H-6a,3), 3.87 (m, 1H, H-4'), 3.72 (dd, 1H, H-6a'), 3.57 (m, 2H, H-6b, 6b'), 3.43 (s, 3H, OCH₃), 3.42 (m, 2H, H-5, 5'), 2.90 (t, $J = 5.0$ Hz, 2H, CH₂N₃), 1.92 (s, 3H, CH₃CON) and 1.90 (s, 3H, Ac).

Anal. Calcd for C₅₆H₆₆O₁₄N₄ (1019.16): C, 66.00; H, 6.53; N, 5.50. Found: C, 65.37; H, 6.68; N, 5.65.

5-Azido-3-oxapentyl 2-Acetamido-6-O-benzyl-4-O-(2-O-acetyl-3,4,6-tri-O-benzyl-β-D-galactopyranosyl)-2-deoxy-β-D-glucopyranoside (12). a) A solution of **11** (100 mg, 0.098 mmol) and DDQ (46 mg, 0.19 mmol) in dichloromethane/water 50:1 (4 mL) was stirred for 20 min at rt. Dichloromethane (5 mL) was added and the solution was washed with saturated aqueous sodium hydrogen carbonate (3 mL), water (3 mL), dried and concentrated. Column chromatography (dichloromethane/acetone 6:1 v/v) of

the residue afforded **12** as a syrup (74 mg, 83%); **b**) **10** (214 mg, 0.21 mmol) was treated with DDQ (150 mg, 0.63 mmol) in dichloromethane/water 50:1 (4 mL) for 5 h at rt. Workup and column chromatography as described in (a) afforded **12** (80 mg, 42%); $[\alpha]_D +36.6^\circ$ (*c* 0.6, chloroform); R_F 0.30 (dichloromethane/acetone 4:1 v/v); 1H NMR ($CDCl_3$) δ 7.45-7.37 (m, 20H, Ph), 5.77 (d, *J* = 8.2 Hz, 1H, NH), 5.32 (dd, $J_{1,2} = 7.8$ Hz, $J_{2,3} = 9.8$ Hz, 1H, H-2'), 4.67 (d, $J_{1,2} = 6.2$ Hz, 1H, H-1), 4.35 (d, 1H, H-1'), 4.00-3.30 (m, 13H, H-2, 3, 3', 4, 4', 5, 5', 6a, 6b, 6', 6'b, CH_2N_3), 1.99 (s, 3H, CH_3CON) and 1.97 (s, 3H, Ac).

Anal. Calcd for $C_{48}H_{58}O_{13}N_4$ (899.01): C, 64.13; H, 6.50; N, 6.23. Found: C, 63.65; H, 6.68; N, 6.09.

5-Azido-3-oxapentyl 2-Acetamido-4-O-(3,4,6-tri-O-benzyl- β -D-galactopyranosyl)-6-O-benzyl-2-deoxy- β -D-glucopyranoside (13). To a solution of compound **12** (70 mg, 0.078 mmol) in dry methanol (1 mL), was added a 0.1 M methanolic solution of sodium methoxide (0.1 mL). After 16 h, the reaction was neutralized with Dowex-50 (H^+) resin, filtered and concentrated to afford **13** (64 mg, 96%) as a syrup: $[\alpha]_D +18.4^\circ$ (*c* 2.24, chloroform); R_F 0.45 (dichloromethane/acetone 4:1 v/v); 1H NMR ($CDCl_3$) δ 7.45-7.37 (m, 20H, Ph), 6.18 (d, *J* = 7.7 Hz, 1H, NH), 4.69 (d, $J_{1,2} = 7.4$ Hz, 1H, H-1), 4.26 (d, $J_{1,2'} = 7.8$ Hz, 1H, H-1'), 3.91 (m, 1H, H-2'), 3.82 (m, 1H, H-3), 3.75 (m, 1H, H-4'), 3.68 (m, 1H, H-2), 3.54 (m, 3H, H-5,5',4), 3.34 (dd, $J_{2',3'} = 10$ Hz, $J_{3',4'} = 3$ Hz, 1H, H-3'), 3.30 (t, *J* = 5 Hz, 2H, CH_2N_3) and 1.90 (s, 3H, CH_3CON).

Anal. Calcd for $C_{46}H_{56}O_{12}N_4$ (856.97): C, 64.47; H, 6.59; N, 6.54. Found: C, 64.65; H, 6.78; N, 5.37.

5-Azido-3-oxapentyl 2-Acetamido-4-O-(2-O-acetyl-3,4,6-tri-O-benzyl- β -D-galactopyranosyl)-3-O-(2,3,4-tri-O-benzyl- α -L-fucopyranosyl)-6-O-benzyl-2-deoxy- β -D-glucopyranoside (14). A solution of **12** (28 mg, 31.7 μ mol), tetraethylammonium bromide (7 mg, 32 μ mol) in dichloromethane (0.5 mL) containing molecular sieves (100 mg) was stirred for 15 min at rt and then a solution of 2,3,4-tri-*O*-tribenzyl- α -L-fucosyl bromide (41 mg, 82 μ mol) in dichloromethane (1 mL) was added. After being stirred for 2 d at rt, the mixture was diluted with dichloromethane (5 mL) and filtered through Celite. The filtrate was washed with water (3 mL), saturated aqueous sodium hydrogen carbonate (3 mL) and water (3 mL), then dried and concentrated. Column chromatography (dichloromethane/acetone 10:1 v/v) of the residue afforded **14** (34 mg,

82 %) as a syrup; $[\alpha]_D -93.0^\circ$ (*c* 0.64, chloroform); R_F 0.50 (dichloromethane/acetone 8:1 v/v); 1H NMR ($CDCl_3$) δ 7.45-7.28 (m, 50H, Ph), 5.88 (d, *J* = 6.9 Hz, 1H, NH), 5.02 (d, $J_{1,2}$ = 3.6 Hz, 1H, H-1f), 4.90 (d, *J* = 3 Hz, 1H, H-1), 4.47 (d, 1H, *J* = 6 Hz, H-1'), 4.37 (m, 1H, H-5f), 3.98 (m, 2H, H-4', 2f), 3.86 (m, 2H, H-3, 3f), 3.74 (m, 2H, 6'a, 6a), 3.60 (m, 3H, H-4, 6b, 6'b), 3.50 (m, 2H, H-2, 5'), 3.30 (m, 5H, 5, 3', 4f, CH_2N_3), 1.75 (s, 3H, Ac) and 1.09 (d, *J* = 6.3 Hz, 3H, C-6f).

Anal. Calcd for $C_{75}H_{86}O_{17}N_4$ (1315.52): C, 68.48; H, 5.59. N, 4.26. Found: C, 68.65; H, 6.86; N, 4.93.

5-Azido-3-oxapentyl 2-Acetamido-4-*O*-(3,4,6-tri-*O*-benzyl-2-*O*-[2,3,4-tri-*O*-benzyl- α -L-fucopyranosyl]- β -D-galactopyranosyl)-3-*O*-(2,3,4-tri-*O*-benzyl- α -L-fucopyranosyl)-6-*O*-benzyl-2-deoxy- β -D-glucopyranoside (15). A solution of 13 (70 mg, 81.7 μ mol), tetraethylammonium bromide (3.5 mg, 16.9 μ mol) in dichloromethane (1 mL) containing molecular sieves (200 mg) was stirred for 15 min at rt and then a solution of 2,3,4-tri-*O*-tribenzyl- α -L-fucosyl bromide (41 mg, 82 μ mol) in dichloromethane (1 mL) was added. After being stirred for 2 d at rt, the mixture was diluted with dichloromethane (5 mL) and filtered through Celite. The filtrate was washed with water (3 mL), saturated aqueous sodium hydrogen carbonate (3 mL) and water (3 mL), then dried and concentrated. Column chromatography (hexane/ ethyl acetate 1:1 v/v) of the residue afforded 15 as a syrup (80 mg, 58%); $[\alpha]_D -48.0^\circ$ (*c* 1.56, chloroform); R_F 0.35 (hexane/ethyl acetate 1:1 v/v); 1H NMR ($CDCl_3$) δ 7.45-7.28 (m, 50H, Ph), 5.77 (d, *J* = 6.9 Hz, 1H, NH), 5.67 (d, $J_{1,2}$ = 3.4 Hz, 1H, H-1f'), 5.15 (d, $J_{1,2}$ = 8.6 Hz, 1H, H-1), 4.91 (d, *J* = 3 Hz, 1H, H-1f), 4.58 (q, 1H, H-5f), 4.47 (d, *J* = 7.7 Hz, 1H, H-1'), 4.20 (m, 2H, H-3, 5f'), 4.00 (m, 5H, 4', 2f', 2f, 2', 4), 3.83 (m, 3H, 6a, 3f, 3f'), 3.67 (m, 4H, 6'a, 4f', 6b, 6'b), 3.53 (dd, $J_{2,3}$ = 8.2 Hz, $J_{3,4}$ = 2.4 Hz, 1H, 3'), 3.35 (t, 2H, *J* = 5 Hz, CH_2N_3), 3.25 (m, 3H, 5', 5, 4f), 1.97 (s, 3H, Ac); 1.28 (d, 3H, H-6f') and 1.10 (d, 3H, H-6f).

Anal. Calcd for $C_{100}H_{112}O_{20}N_4$ (1689.00): C, 71.11; H, 6.62. N, 3.32 Found: C, 70.62; H, 6.81, N, 3.20.

5-Amino-3-oxapentyl 2-Acetamido-2-deoxy-3-*O*-(α -L-fucopyranosyl)-4-*O*-(2-*O*-[α -L-fucopyranosyl]- β -D-galactopyranosyl)- β -D-glucopyranoside (16). A solution of compound 15 (78 mg, 0.046 mmol) in ethyl acetate/methanol/water/acetic acid 5:5:1:0.1 v/v (1 mL) containing 10 % palladium on carbon (40 mg) was stirred overnight

under H₂ at rt. The mixture was filtered and the solid washed with water (25 mL). The filtrate and washings were concentrated under reduced pressure and lyophilized to afford **16** as a colorless powder (32 mg, 90%); $[\alpha]_D -102^\circ$ (*c* 0.27, water); R_F 0.23 (ethyl acetate/ methanol/ water/ acetic acid 5:5:1:0.1 v/v); ¹H NMR (D₂O) δ 5.32 (d, J_{1,2} = 2.5 Hz, 1H, H-1f'), 5.16 (d, J_{1,2} = 3.9 Hz, 1H, H-1f), 4.91 (q, 1H, H-5f), 4.66(d, J_{1,2} = 8.3 Hz 1H, H-1'), 4.57(d, J=7.8 Hz, 1H, H-1), 4.32 (q, 1H, H-5f'), 4.10 (dd, J_{6,6a} = 10.5 Hz, J_{5,6a} = 2.4 Hz, 1H,6a), 3.99 (m, 2H, H-4, 3f), 3.91 (m, 9H, H-4', 3', 3, 2, 6b, 3f', 4f', 4f, 2f'), 3.80 (m, 3H, H-6'a, 6'b, 2f), 3.72 (dd, J_{2,3} = 8.5 Hz, 1H, H-2), 3.56(m, 2H, H-5', 5), 3.14 (t, 2H, CH₂NH₂), 2.10 (s, 3H, CH₃CON), 1.36 (d, 3H, H-6f') and 1.33 (d, 3H, H-6f); ¹³C NMR (CDCl₃) δ 170.5 (C=O), 101.8 (C-1), 101.1 (C-1'), 100.2 (C-1f'), 99.2 (C-1f), 77.3 (C-2'), 76.5 (C-3), 75.8 (C-5), 75.7 (C-5'), 74.6 (C-4), 74.4 (C-3'), 73.0 (C-4f), 72.6 (C-4f'), 70.4 (C-3f'), 69.8 (C-3f), 69.5 (C-4'), 69.0 (C-2f), 68.8 (C-2f'), 67.5 (C-5f, 5f'), 62.1 (C-6'), 60.9 (C-6), 56.8 (C-2), 40.3 (CH₂NH₂), 23.4(CH₃CON) and 16.4 (C-6f, 6f').

Anal. Calcd for C₃₀H₃₄O₂₀N₂ (761.65): C, 47.24; H, 7.14; N, 3.67. Found: C, 47.36; H, 7.20; N, 3.78.

5-Amino-3-oxapentyl 2-Acetamido-2-deoxy-3-O-(α-L-fucopyranosyl)-4-O-(β-D-galactopyranosyl)-β-D-glucopyranoside (17). A solution of compound **14** (20mg, 0.014 mmol) was dissolved in a dry solution of methanolic sodium methoxide 0.01 M (1 mL) and stirred for 16 h at rt. The solution was then neutralised with Amberlite IR 120 (H⁺) resin. The resulting mixture was filtered and the filtrate concentrated to a syrupy residue that was hydrogenolised over 10 % palladium on carbon (15 mg) in ethyl acetate/ methanol/water/acetic acid 5:5:1:0.1 v/v (0.5 mL) overnight at rt. The resulting mixture was filtered and the solid washed with water (20 mL). The filtrate was concentrated under reduced pressure and lyophilized to afford **17** as a colorless powder (8.2 mg, 89%); R_F 0.29 (ethyl acetate/methanol/water/acetic acid 5:5:1:0.1 v/v); ¹H NMR (D₂O) δ 5.18 (d, J_{1,2} = 3.9 Hz, 1H, H-1f), 4.91 (q, 1H, H-5f), 4.59(d, J_{1,2} = 8.3 Hz 1H, H-1'), 4.55 (d, J=7.8 Hz, 1H, H-1), 4.09 (dd, J_{6,6'} = 10.5 Hz, J_{5,6} = 2.4 Hz, 1H, H-6), 3.99 (m, 2H, H-4, 3f), 3.91 (m, 9H, H-4', 3', 3, 2, 6a, 4f), 3.80 (m, 3H, H-6b, 6'a, 2f), 3.56(m, 3H, H-5', 2', 5), 3.14 (t, 2H, CH₂NH₂), 2.12 (s, 3H, CH₃CON) and 1.34 (d, 3H, H-6f); ¹³C NMR (CDCl₃) δ 170.6 (C=O), 101.9 (C-1), 101.0 (C-1'), 99.5 (C-1f), 76.7 (C-3), 75.8 (C-5), 75.6 (C-5'), 74.4 (C-3'), 74.6 (C-4), 73.0 (C-4f), 71.3 (C-2'), 69.8 (C-3f), 69.5 (C-4'), 69.0 (C-2f), 67.3 (C-5f), 61.8 (C-6'), 61.2 (C-6), 55.8 (C-2), 40.2 (CH₂NH₂), 23.4 (CH₃CON) and 16.3 (C-6f).

Anal. Calcd for C₂₄H₄₄O₁₇N₂(632.62): C, 45.57; H, 7.01. N, 4.43 Found: C, 45.01; H, 7.45, N, 4.35.

Coupling reaction between 16 or 17 and BSA. To a solution of the free oligosaccharide 16 or 17 (4 μmol) in *N,N*-dimethylformamide (distilled in *vacuo* and pumped for 30 min before use, 0.5 mL) was added the *N*-hydroxysuccinimide derivative of β-maleimidopropanoic acid²⁵ (1.3 mg, 5 μmol). After 2 h, the solution was concentrated and dried *in vacuo*. The residue was resuspended in D₂O (0.5 mL) and centrifuged to remove the excess of reagent. ¹H NMR (D₂O) δ 6.89 (s, 2H, HC=CH), 3.42 (t, 2H, CH₂N), 2.52 (t, 2H, CH₂CO). The maleimido derivative 18 or 19 could be stored in a lyophilised form at 0°C or used directly in the coupling reaction.

Compound 18 or 19 was added to a solution²⁶ of BSA-SH₂₃ in PBS (pH 7.2, 0.4 mL). After 2 h, the process was complete as evidence by negative Ellman test²⁷ and the resulting solution was dialyzed against PBS (pH = 7.4). The protein and carbohydrate contents were determined by the Lowry²⁸ and phenol-sulfuric acid methods,²⁹ respectively. Several assays gave carbohydrate to protein ratios between 13-20 oligosaccharide units per BSA molecule.

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REFERENCES

1. Presented at the *VII European Carbohydrate Symposium*, Cracow, Poland, August 22-27, 1993.
2. D. R. Bundle, M. A. J. Gidney, N. Kassam and A. F. R. Rahman, *J. Immunol.*, **129**, 678 (1982).
3. S. Martensson, T. Brodin, A. S. Carlstrom, J. Dahmen, T. Frejd, A. Gunnarsson, U. Jansson, G. Magnusson and A. Lundblad, *Glycoconjugate J.*, **3**, 163 (1986).
4. G. D. McLean, M. Reddish, R. R. Koganty, T. Wong, S. Gandhi, M. Smolenski, J. Samuel, J. M. Nabholt and B. M. Longenecker, *J. Immunol. Immunother.*, **36**, 215 (1993).
5. C. C. A. M. Peeters, D. Evenberg, P. Hoogerhout, H. Kayhty, L. Saarinen, C. A. A. Boeckel, G. A. Marel, J. H. van Boom and J. T. Poolman, *Infect. Immun.*, **60**, 1826 (1992).

6. V. Fernandez Santana, R. Gonzalez Lio, J. Sarracent Perez and V. Verez Bencomo, *VII European Carbohydrate Symposium*, Cracow, Poland, August 22-27, 1993, A-168.
7. P. Avram-Zollo and P. Sinaÿ, *Carbohydr. Res.*, **150**, 199 (1986).
8. S. Nilsson, M. Bengtsson and T. Norberg, *J. Carbohydr. Chem.*, **10**, 1 (1991).
9. N. Nakajima, R. Abe and O. Yonemitsu, *Chem. Pharm. Bull.*, **36**, 4244 (1988).
10. a) H. Lönn, *Carbohydr. Res.*, **139**, 115 (1985); b) S. Sato, Y. Ito, T. Nukuda, Y. Nakahara and T. Ogawa, *Carbohydr. Res.*, **167**, 197 (1987); c) S. Nilsson, H. Lönn and T. Norberg, *Glycoconjugate J.*, **6**, 21 (1989); d) R. R. Schmidt and D. Toepfer, *Tetrahedron Lett.*, **32**, 3353 (1991); e) W. Kinzy and A. Low, *Carbohydr. Res.*, **245**, 193 (1993).
11. a) J. C. Jacquinet and P. Sinaÿ, *J. Chem. Soc., Perkin Trans. 1*, 314 (1979); b) R. Bommer, W. Kinzy and R.R. Schmidt, *Liebigs Ann. Chem.*, 425 (1991); c) U. Spohr and R. U. Lemieux, *Carbohydr. Res.*, **174**, 211 (1988); d) J. C. Jacquinet and P. Sinaÿ, *J. Org. Chem.*, **42**, 720 (1977); e) O. Hindsgaul, T. Norberg, J. LePendu and R. U. Lemieux, *Carbohydr. Res.*, **109**, 104 (1982).
12. S. Sabesan and R. U. Lemieux, *Can. J. Chem.*, **62**, 135 (1984); R. R. Schmidt and G. Grundler, *Liebigs Ann. Chem.*, 1826 (1984).
13. M. Nilsson and T. Norberg, *Carbohydr. Res.*, **183**, 71 (1988).
14. M. Dejter-Juzynsky and H. M. Flowers, *Carbohydr. Res.*, **18**, 219 (1971).
15. K. Heyns, R. Harrison and H. Paulsen, *Chem. Ber.* **100**, 271 (1967).
16. P. J. Garegg, H. Hultberg and S. Wallin, *Carbohydr. Res.*, **108**, 97 (1982).
17. S. Hanessian and J. Banoub in *Methods Carbohydr. Chem.*, Vol VIII, R. L. Whistler and J. N. BeMiller, Eds.; Academic Press: New York, 1980, p 247.
18. M. Nilsson and T. Norberg, *Carbohydr. Res.*, **183**, 71 (1988).
19. M. Dejter-Juzynsky and H. M. Flowers, *Carbohydr. Res.*, **18**, 219 (1971).
20. T. Iversen and D. Bundle, *Carbohydr. Res.*, **103**, 29 (1982).
21. R. U. Lemieux, K. B. Hendricks, R. V. Stick and K. James, *J. Am. Chem. Soc.*, **97**, 4056 (1975).
22. V. Fernandez Santana, R. Gonzales Lio, J. Sarracent Pérez and V. Verez Bencomo., *Glycoconjugate J.*, in press.
23. S. Nilsson, M. Bengtsson and T. Norberg, *J. Carbohydr. Chem.*, **10**, 1 (1991).
24. D. Horton in *Methods Carbohydr. Chem.*, Vol VI, R. L. Whistler and J. N. BeMiller, Eds.; Academic Press: New York, 1972, p 282.
25. O. Keller and J. Rudinger, *Helv. Chim. Acta*, **58**, 531 (1975).
26. V. Fernandez Santana, R. Gonzalez Lio, J. Sarracent Perez and V. Verez Bencomo, *Glycoconjugate J.*, submitted.
27. G. L. Ellman, *Arch. Biochem. Biophys.*, **74**, 443 (1958).
28. C. Capla, *Methods Enzymol.*, **73**, 448 (1957).
29. M. Dubois, K. A. Gilles, J. K. Hamilton, P. A. Rebers and F. Smith, *Anal. Biochem.*, **28**, 350 (1956).